

CM3215

**MichiganTech**

Fundamentals of Chemical Engineering Laboratory

**Overall Heat Transfer Coefficient for Double-Pipe Heat Exchanger**

**Professor Faith Morrison**

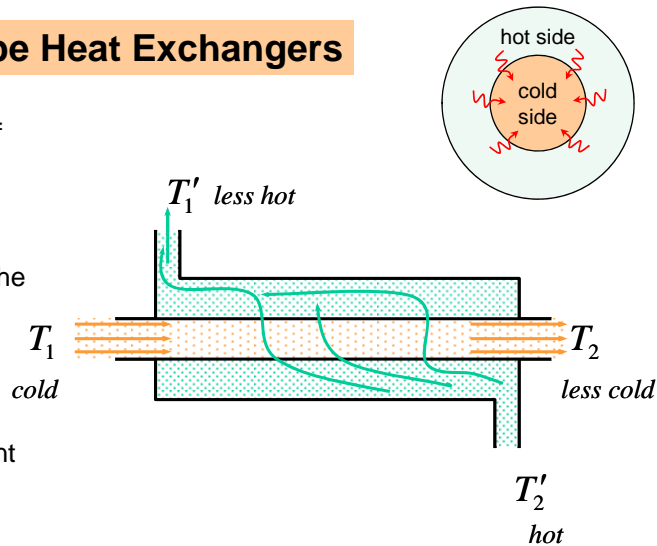
Department of Chemical Engineering  
Michigan Technological University

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**Double-Pipe Heat Exchangers**

- Simplest type of heat exchanger
- Total heat transferred proportional to the log-mean temperature difference
- Counter-current shown; co-current also possible



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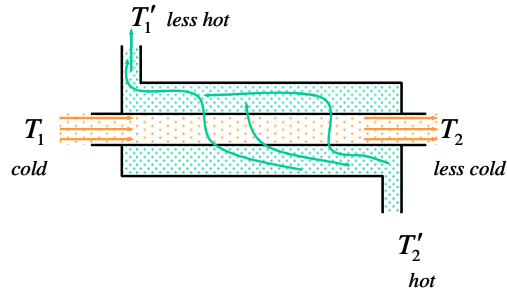
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## Double-Pipe Heat Exchangers

Steady state macroscopic energy balance on:

- Inside
- Outside
- Overall

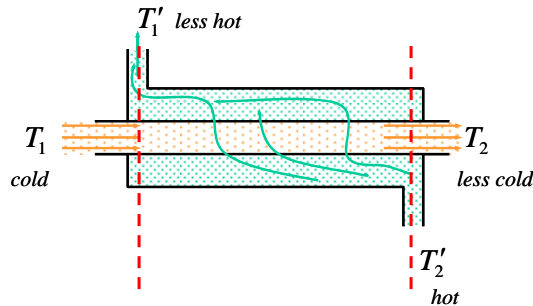
Covered in CM3110 Transport Processes I



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## Double-Pipe Heat Exchangers



$$\Delta T_{left} = (T_1' - T_1)$$

$$\Delta T_{right} = (T_2' - T_2)$$

- $\Delta T = \text{driving force for heat transfer}$
- $\Delta T$  varies down the length of the heat exchanger (HE)
- Amount of heat transferred varies down the length of the HE

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### Double-Pipe Heat Exchangers

- $\Delta T = \text{driving force for heat transfer}$
- $\Delta T$  varies down the length of the heat exchanger (HE)
- Amount of heat transferred varies down the length of the HE

The overall heat transferred  $Q$  is governed by the average driving force,  $\Delta T_{average}$

The correct average driving force (as we show in CM3110) is the **log mean  $\Delta T$**

*Comes from considering the energy balances)*

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### Log Mean $\Delta T$ Driving Force

The overall heat transferred is governed by the **log mean driving force**

Driving temperature difference at the left side of heat exchanger

Driving temperature difference at the right side of heat exchanger

$$\Delta T_{lm} \equiv \frac{(T_1' - T_1) - (T_2' - T_2)}{\ln \frac{(T_1' - T_1)}{(T_2' - T_2)}}$$

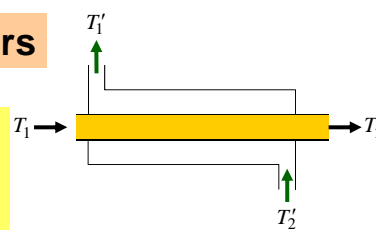
Note: the log-mean average temperature driving force will be a number between these two  $\Delta T$ 's

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## Double-Pipe Heat Exchangers

- Total Heat Transferred given by  $Q$
- The HE is characterized by its overall heat transfer coefficient,  $U$



$$Q = UA\Delta T_{lm}$$

$$= U(2\pi R_{outer}L)\Delta T_{lm}$$

$$U \equiv \frac{Q}{A\Delta T_{lm}}$$

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### Where do we get $A$ ?

From the Equipment Schedule  
(see website):

8/17/2001 DWG		Department of Chemical Engineering Michigan Tech University				
Equipment Schedule for Fundamentals of CM Laboratory Apparatus						
Tag	Location of Device	Comments	Product	Flowrate	Specific Gravity	Temp., deg. F
E-01	Above branched piping circuit	Double-pipe Heat Exchanger, Shell-2" Sched. 40 C.S. pipe, Tube 3/8" o.d. type L copper tubing, 48" long, 3 gpm @ 71 ft. head, 1 1/4" x 1", 120/240	City Water/ Steam	5.12 GPM	1	55
P-01	Below bench	VAC, 1ph., 1/2 hp open drip-proof motor	City Water	5 GPM	1	55
T-01	Below W007 outlet	Tank, 10 gallon, heavy wall PE, 18 3/4" h x 13" o.d.	City Water	5 GPM	1	55
T-02	Below W008 outlet	Tank, 10 gallon, heavy wall PE, 18 3/4" h x 13" o.d.	City Water	5 GPM	1	55

Key to Miscellaneous Identifiers	
PT-##	Pressure Tap
T#	Steam Trap

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We will record data (and first calculations) on the white board in class – be prepared to do so

- Temperatures
- $\Delta T_{lm}$
- Heat flow (steam side)
- Heat flow (water side)
- $U \left( \frac{W}{m^2K} \right)$ , water side
- $U \left( \frac{W}{m^2K} \right)$ , steam side

Answer the SurveyMonkey (or Google forms) to submit your data – be prompt!

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How do we measure  $Q$ ?

We will record data (and first calculations) on the white board in class – be prepared to do so

- |  |  |
|--|--|
| <ul style="list-style-type: none"> <li>• Temperatures</li> <li>• <math>\Delta T_{lm}</math></li> <li>• Heat flow (steam side)</li> <li>• Heat flow (water side)</li> <li>• <math>U \left( \frac{W}{m^2K} \right)</math>, water side</li> <li>• <math>U \left( \frac{W}{m^2K} \right)</math>, steam side</li> </ul> | <p>Two ways:</p> <ul style="list-style-type: none"> <li>• Water side</li> <li>• Steam side</li> <li>• Perform <i>Macroscopic Energy Balances</i> to obtain <math>Q</math></li> </ul> |
|--|--|

Answer the SurveyMonkey (or Google forms) to submit your data – be prompt!

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How do we measure  $Q$ ?

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|---|--|
| <ul style="list-style-type: none"> <li>• Temperatures</li> <li>• <math>\Delta T_{lm}</math></li> <li>• Heat flow (steam)</li> <li>• Heat flow (water)</li> <li>• <math>U \left( \frac{W}{m^2K} \right)</math>, water</li> <li>• <math>U \left( \frac{W}{m^2K} \right)</math>, steam side</li> </ul> | <p>Two ways:</p> <ul style="list-style-type: none"> <li>• Water side</li> <li>• Steam side</li> <li>• Perform <i>Macroscopic Energy Balances</i> to obtain <math>Q</math></li> </ul> |
|---|--|

**Macro-E-Balance comes from CM2110 and CM2120;  
see our handout page for Dr. Morrison's E- Balance**

**Notes:** [www.chem.mtu.edu/~fmorriso/cm310/Energy\\_Balance\\_Notes\\_2008.pdf](http://www.chem.mtu.edu/~fmorriso/cm310/Energy_Balance_Notes_2008.pdf)

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## CM3215 Fundamentals of Chemical Engineering Laboratory

### Report 6: Overall Heat Transfer Coefficient for Double-Pipe Heat Exchanger

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- Pump water through inner pipe of double-pipe heat exchanger
- Flow steam through outer pipe of double-pipe heat exchanger
- Allow to come to steady state (monitor; prove steady state)
- Measure appropriate temperatures, flow rates and replicates
- Share raw data with classmates; use all data from your lab station across 5 lab sections
- Calculate and report **overall heat transfer coefficient**; use two different methods
- Report the process of achieving steady state
- Address all objectives in assignment memo

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Pneumatic Valve Position

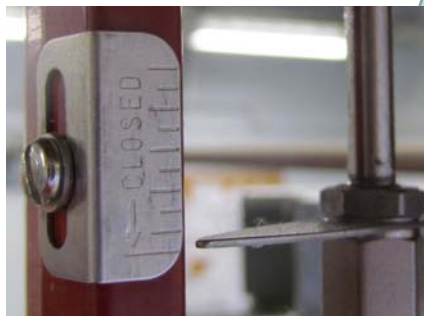


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Note: The position of the valve stem is read from a set of notches based on 1/8<sup>th</sup> distance per notch:

(compare to a 12 inch ruler)

x= fraction open



(5.5)/8ths open

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**CM3215 Fundamentals of Chemical Engineering Laboratory****Prelab Assignment:**

- Review the following: concept of the double-pipe heat exchanger, the overall heat transfer coefficient, logarithmic mean temperature difference and purpose and operation of steam traps.
- Calculate the heat transfer area of the double pipe heat exchanger in the lab. Base your calculation on the outside area of the inside pipe.
- Fit the latent heat of vaporization of steam over a reasonable temperature range (near atmospheric pressure) to an empirical function
- Be prepared to make some trial calculations
- Prepare a safety section

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