

CM 3230 Fall 2011

Quiz 6a

Name \_\_\_\_\_

Circle the correct answers, each question is worth 20 points. (Bonus of 20 if all 6 are correct).

1. For a liquid solution having 30% A and 70% B, the activity coefficients were found to be  $\gamma_A = 1.2$  and  $\gamma_B = 2$ . Then,

- a)  $g^E/(RT) = 0.11$
- b)  $g^E/(RT) = 0.36$
- c)  $g^E/(RT) = 0.54$
- d) None of the above

2. For a binary vapor mixture in equilibrium with a liquid solution, the fugacity of A was found to be  $\hat{f}_1^v = 2$  bars. The liquid composition was found to be  $x_A = 0.5$  with an activity coefficient of 1. Assuming that the  $P^{sat} = 4.3$  bars and  $\phi_1^{sat} = 1$ , then the Poynting correction,  $\epsilon_P$ , is

- a)  $\epsilon_P = 0.72$
- b)  $\epsilon_P = 0.93$
- c)  $\epsilon_P = 1.07$
- d) None of the above

3. A mixture containing 65 mole % A and 35 mole% B at  $T = 300K$  and  $P = 2$  bars was found to have an excess Gibbs free energy  $g^E = -10$  kJ/mol. Then  $\Delta g_{mix}$  is

- a)  $\Delta g_{mix} = -8.4 \frac{kJ}{mol}$
- b)  $\Delta g_{mix} = -11.6 \frac{kJ}{mol}$
- c)  $\Delta g_{mix} = -30.2 \frac{kJ}{mol}$
- d) None of the above

4. A binary liquid mixture containing 15%  $A$  and 85%  $B$  is in equilibrium with the vapor mixture containing 20%  $A$  and 80%  $B$  at a pressure  $P = 2$  bars. If  $f_A^l = 1$  bar with an activity coefficient  $\gamma_A = 1.6$ , then the fugacity coefficient of  $A$  in the vapor is given by
- $\hat{\phi}_A = 0.3$
  - $\hat{\phi}_A = 0.6$
  - $\hat{\phi}_A = 1.2$
  - None of the above
5. For a binary mixture at  $T = 400K$  and  $P = 1.5$  bar, the activity coefficient  $\gamma_A$  at infinite dilution, i.e. near  $x_B = 1$ , was found to be 1.5. Assuming that the Margules 2-suffix model,  $g^E = A x_A x_B$  is appropriate for the solution, then we have
- $\frac{A}{R} = 142 K$
  - $\frac{A}{R} = 212 K$
  - $\frac{A}{R} = 420 K$
  - None of the above
6. From the generalized correlation, the following values were found for pure  $A$ :  $\ln(\phi^{(0)}) = -0.3$ ,  $\ln(\phi^{(1)}) = 0.1$  and  $\omega = 0.2$  for substance  $A$  at  $P = 200$  bars. Then the fugacity is given by
- $f_A = 122$  bars
  - $f_A = 203$  bars
  - $f_A = 151$  bars
  - None of the above